

Al-Co-Si (Aluminum-Cobalt-Silicon)

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The previous reports on this system are by [1951Pra] (partial isothermal section at 536 °C for Al-rich alloys) and by [1981Her] (full isothermal section at 600 °C). Recently, [2005Ric] investigated the phase equilibria and reported three isothermal sections at 800 and 600 °C for Co-lean alloys and at 900 °C for Co-rich alloys.

Binary Systems

The Al-Co phase diagram [1996God] shows the following intermediate phases: CoAl (48–78.5 at.% Co; *B*2, CsCl-type cubic), Co₂Al₅ (*D*8₁₁-type hexagonal), CoAl₃ (*D*0₁₁, Fe₃C-type orthorhombic), Co₄Al₁₃ (three modifications with one orthorhombic and two monoclinic forms), and Co₂Al₉

(*D*8_d-type monoclinic). The Al-Si phase diagram is a simple eutectic system with the eutectic reaction at 577 °C and 12.2 at.% Si. The Co-Si phase diagram [1991Ish] depicts the following intermediate phases: Co₃Si (tetragonal, stable between 1214 and 1193 °C), βCo₂Si (stable between 1334 and 1238 °C), αCo₂Si (*C*23, Co₂Si-type orthorhombic), CoSi (*B*20, FeSi-type cubic), and CoSi₂ (*C*1, CaF₂-type cubic). The (αCo) → (εCo) transition temperature is steeply raised to above 1200 °C by the addition of Si.

Ternary Phases

[2005Ric] presented the structural data on seven ternary phases of this system, by combining the previously-known

Table 1 Al-Co-Si crystal structure and lattice parameter data [2005Ric]

Phase	Composition, at.%	Pearson symbol	Space group	Prototype	Parameter Lattice, nm
Co ₃ Al _{3.1-2.4} Si _{3.9-4.6} (τ ₁ or α)	31-24 Al 30 Co 39-46 Si	<i>c</i> 140	<i>Im</i> ̄3̄ <i>m</i>	Ge ₇ Ir ₃	<i>a</i> = 0.8115 – 0.8072
Co ₄ Al _{1.7-2.8} Si _{4.3-3.2} (a) (τ ₂ or β)	17-28 Al 40 Co 43-32 Si	<i>h</i> P5	<i>P</i> ̄3̄ <i>m</i> 1	Ni ₂ Al ₃	<i>a</i> = 0.3870 – 0.3885 <i>c</i> = 0.4758 – 0.4769
Co _{19+x} Al _{43+y} Si _{12-y} (0 ≤ <i>x</i> ≤ 0.7; -0.5 ≤ <i>y</i> ≤ 3) (τ ₃ or γ)	56.9-62.2 Al 25.7-26.4 Co 16.9-12.2 Si	<i>m</i> C296+	<i>C</i> 2/ <i>c</i>	...	<i>a</i> = 2.0040 <i>b</i> = 1.9170 <i>c</i> = 1.2826 γ = 123.591°
Co _{32.5} Al _{40.5-43} Si _{27-24.5} (τ ₄ or δ)	40.5-43 Al 32.5 Co 27-24.5 Si	<i>m</i> C?	<i>a</i> = 1.1851 <i>b</i> = 0.38838 <i>c</i> = 0.74293 γ = 103.17°
Co ₆ Al ₁₁ Si ₆ (τ ₅ or ε)	47.8 Al 26.1 Co 26.1 Si	<i>o</i> C184	<i>Cmc</i> 2 ₁	...	<i>a</i> = 0.80839 <i>b</i> = 1.45445 <i>c</i> = 2.1354
Co ₄ Al _{7+x} Si _{2-x} (0.27 ≤ <i>x</i> ≤ 1.05) (τ ₆ or χ)	55.9-61.9 Al 30.8 Co 13.3-7.3 Si	<i>m</i> C26	<i>C</i> 2/ <i>m</i>	...	<i>a</i> = 1.19935 <i>b</i> = 0.40098 <i>c</i> = 0.76590 γ = 106.357°
Co _{10+x} Al _{25-x} Si ₇ (<i>x</i> = 0.33) (τ ₇ or φ)	58.7 Al 24.6 Co 16.7 Si	<i>o</i> P168	<i>Pnma</i>	...	<i>a</i> = 1.3846 <i>b</i> = 2.3050 <i>c</i> = 0.7336

(a) Co variation not indicated (see text)

Section II: Phase Diagram Evaluations

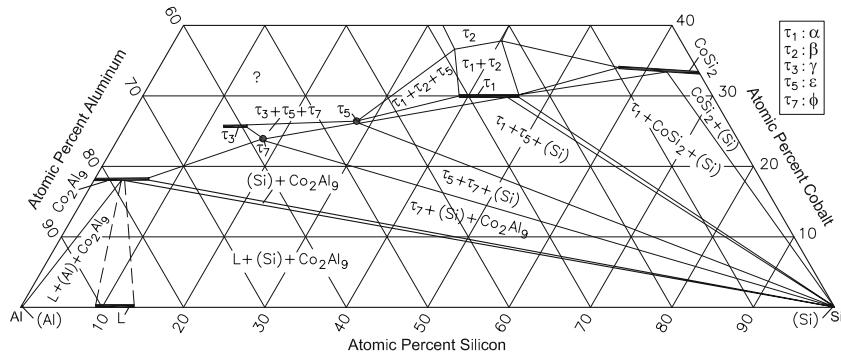


Fig. 1 Al-Co-Si partial isothermal section at 600 °C [2005Ric]. Narrow two-phase regions are omitted

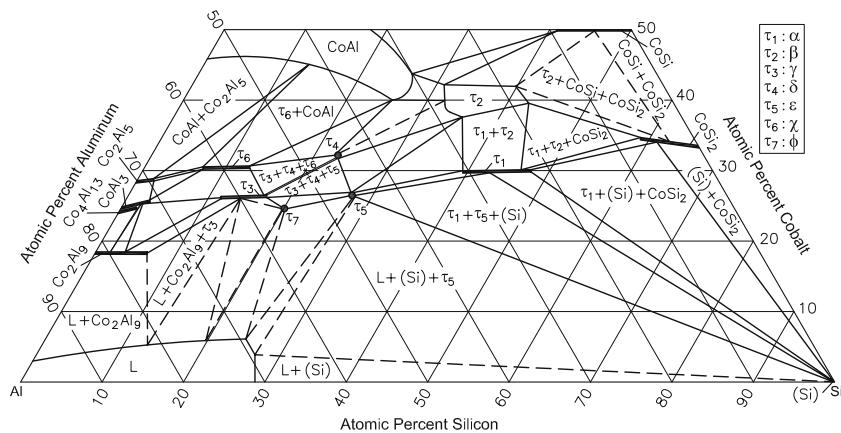


Fig. 2 Al-Co-Si isothermal section at 800 °C [2005Ric]. Narrow two-phase regions are omitted

data with their own work, see Table 1. The ternary phases are denoted here as τ_1 , τ_2 , τ_3 , τ_4 , τ_5 , τ_6 , and τ_7 . They correspond to α , β , γ , δ , ϵ , χ , and ϕ , respectively in the nomenclature adopted by [2005Ric]. The phase τ_2 (β) shows a variation in the Co content of about ± 2 at.% around the ideal value of 40 at.% [2005Ric].

Ternary Isothermal Sections

With starting metals of 99.999% Al, 99.9+% Co and 99.9999% Si, [2005Ric] arc-melted under Ar atm a total of 104 alloy compositions. A series of 28 samples with Co content up to 35 at.% were annealed at 600 °C for 4 weeks. A second series of 60 samples with Co content up to 50 at.% and in the Al-rich region, wherein lie the ternary phases τ_3 , τ_4 , τ_5 , τ_6 , and τ_7 , were annealed at 800 °C for 4 weeks. A third series of 26 samples in the Co-rich region were annealed at 900 °C. The phase equilibria were studied

with x-ray powder diffraction, metallography, and electron probe microanalysis.

Figure 1 shows the partial isothermal section at 600 °C constructed by [2005Ric]. The binary phases CoSi₂ and Co₂Al₉ dissolve up to 9.3 at.% Al and 6.6 at.% Si, respectively. The results of [2005Ric] showed that Si substitutes for Al in Co₂Al₉. Figure 2 is the isothermal section at 800 °C. The CoAl (B2) phase extends far into the ternary region. The ternary phases τ_4 (δ), τ_5 (ϵ), and τ_7 (ϕ) are virtually stoichiometric. The Si ranges of τ_1 (α) and τ_6 (χ) are 39–46 and 7.3–13.3 at.%, respectively (at constant Co content). The homogeneity regions of τ_2 (β) has a more complex spread. Figure 3 shows the isothermal section constructed by [2005Ric] in the Co-rich region. The (ϵ Co) phase stabilized by Si dissolves very little Al. CoSi dissolves up to 11.5 at.% Al, whereas the solubility in Co₂Si is only 2.7 at.%. The CoAl (B2) phase extends up to 73 at.% Co, as compared with its extension up to 61 at.% Co in the Al-Co binary system. [2005Ric] reported detailed measurements of the lattice parameters of the binary and ternary solid solutions.

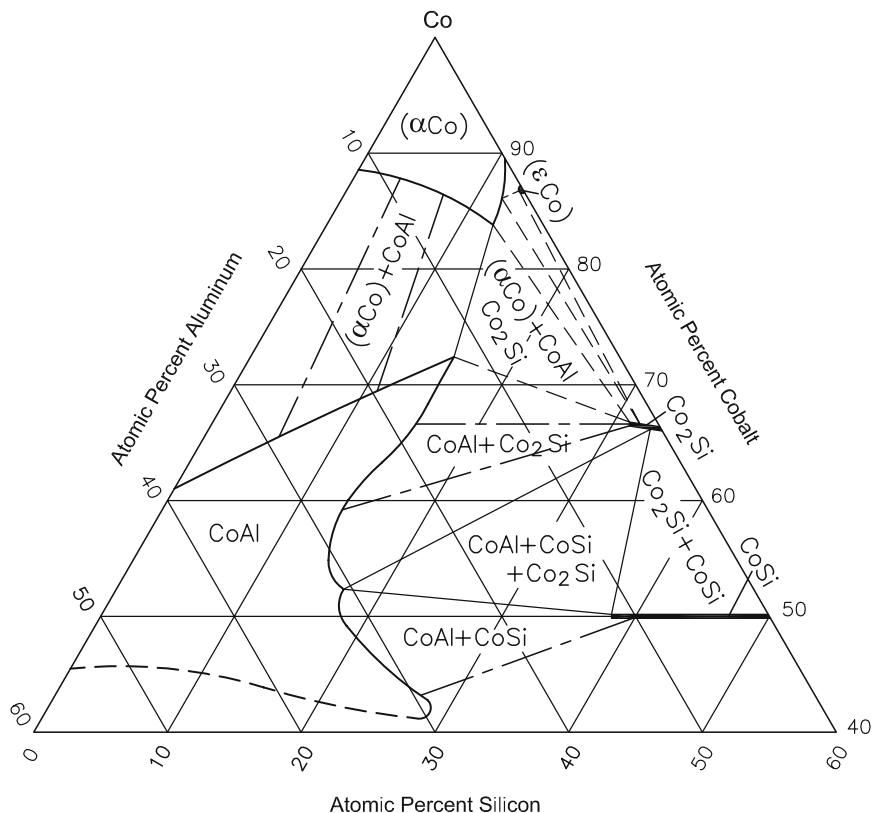


Fig. 3 Al-Co-Si isothermal section at 900 °C [2005Ric]

References

- 1951Pra:** J.N. Pratt and G.V. Raynor, The Intermetallic Compounds in the Alloys of Aluminum and Silicon with Chromium, Manganese, Iron, Cobalt, and Nickel, *J. Inst. Metals (London)*, 1951, **79**, p 211-232
- 1981Her:** N.V. Herman, The Ternary System Nickel-Silicon-Aluminum and Cobalt-Silicon-Aluminum, *Vissn. L'viv. Derzh. Univ., Ser. Khim.*, 1981, **23**, p 61-64

1991Ish: K. Ishida, T. Nishizawa, and M.E. Schlesinger, The Co-Si (Cobalt-Silicon) System, *Bull. Alloy Phase Diagrams*, 1991, **12**(5), p 578-586

1996God: T. Godecke and M. Ellner, Phase Equilibria in the Aluminum-Rich Portion of the Binary System Co-Al and in the Co/Al-Rich Portion of the Ternary System Co-Ni-Al, *Z. Metallkd.*, 1996, **87**(11), p 854-864

2005Ric: K.W. Richter and D.T. Gutierrez, Phase Equilibria in the System Al-Co-Si, *Intermetallics*, 2005, **13**, p 848-856